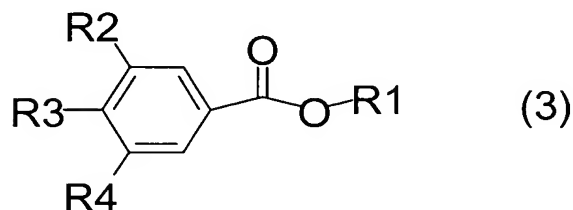


AMENDMENTS TO THE CLAIMS:

This listing of claims will replace all prior versions, and listings, of claims in the application:

LISTING OF CLAIMS:

1. (Currently Amended) A method for preparing a glycoside represented by formula (3):



wherein:

R1 represents a C₁-C₁₀ straight or branched alkyl;

any one of R2, R3 and R4 represents a hydroxyl;

each of the remaining two of R2, R3, and R4 represent a residue formed by reacting glucose pentaacetate with a hydroxyl group of an alkyl gallate; and in which at least two phenolic hydroxyl groups of a compound having at least two phenolic hydroxyl groups in its molecule are glycosidically linked to anomeric carbon atoms of at least two sugar molecules;
the method ~~comprising~~ comprises the step of:

~~allowing a compound having at least two phenolic hydroxyl groups in its molecule to react with sugar molecules each having an acetylated hydroxyl group attached to an anomeric carbon atom~~
reacting an alkyl gallate having a C₁-C₁₀ straight or branched alkyl with glucose pentaacetate in the presence of an organic solvent and an acid catalyst at 30°C to 30°C while removing from the reaction system organic solvent and acetic acid formed during the

reaction, thereby maintaining the concentration of the acetic acid in the reaction system at 1.0 percent by weight or less during the reaction~~from the reaction system.~~

2-20. (Canceled).

21. (Currently Amended) The method for preparing a glycoside according to Claim-20~~1~~, wherein the organic solvent is xylene.

22. (Previously Presented) The method for preparing a glycoside according to Claim 21, wherein the acid catalyst is a Lewis acid catalyst.

23. (Previously Presented) The method for preparing a glycoside according to Claim 21, wherein the acid catalyst is boron trifluoride.

24-27. (Canceled).

28. (New) The method for preparing a glycoside according to Claim 1, wherein in step (b) the organic solvent and the acetic acid are removed from the reaction system by distillation at a rate of 10 to 1000 g/hr relative to one mol of the alkyl gallate.